



Coimisiún na Scrúduithe Stáit
State Examinations Commission

JUNIOR CERTIFICATE EXAMINATION, 2007

SCIENCE (Revised Syllabus)

HIGHER LEVEL

Marking Scheme

TABLE FOR ASSIGNING GRADES

GRADE	RANGE
A	510 - 600
B	420 - 509
C	330 - 419
D	240 - 329
E	150 - 239
F	60 - 149
NG	0 - 59

General Points regarding the Marking Scheme for Junior Certificate Science

1. In many cases only key phrases are given in the marking schemes. These points contain the information and ideas that must appear in the candidate's answer in order to merit the assigned marks.
2. The descriptions, methods and definitions given in a marking scheme are not exhaustive and alternative valid answers are acceptable.
3. The detail required in any answer is determined by the context and the manner in which the question is asked and by the number of marks assigned to the answer in the examination paper. This may vary from year to year.
4. The bold text is often used to indicate the essential points required in the candidate's answer. A double solidus (//) separates points for which separate marks are allocated in a part of the question. Words, expressions or statements separated by a solidus (/) are alternatives which are equally acceptable for a particular point. A word or phrase in bold, given in brackets, is an acceptable alternative to the preceding word or phrase. Note, however, that words, expressions or phrases must be correctly used in context and not contradicted. Where there is evidence of incorrect use or contradiction, the marks may not be awarded.
5. In general, names and formulas of elements and compounds are equally acceptable except in cases where either the name or the formula is specifically asked for in the question. However, in some cases where the name is asked for, the formula may be accepted as an alternative. This is clarified within the scheme.
6. There is a deduction of one mark for each arithmetical slip made by a candidate in a calculation.
7. **Cancelled &/or Repeated Answers**

In the case of short-answer questions, if an answer is cancelled and a second answer given, the cancellation is accepted and marks are awarded for the uncanceled answer. If two answers are given and neither answer is cancelled, the first answer offered only is accepted and marked accordingly. If the only answer offered is cancelled, the cancelling is ignored and the answer marked as normal. However, in MCQ-type questions cancelling of an incorrect and correct answer applies.

For answers to "describe an investigation / an experiment", multiple attempts will be dealt with as follows:

If a candidate answers a question or part of a question once only and then cancels, the cancelling is ignored and the answer marked as normal. If a candidate answers a question or part of a question more than once and then cancels one attempt, the cancelling will be ignored and all the answers, whether cancelled or not, marked as normal. However, only the marks gained in respect to the highest scoring attempt will be counted. Points cannot be "mixed and matched from two attempts". The disallowed marks should be enclosed in square brackets.

- 8. Deduction of marks for omitted labelled diagrams**
Assign marks in the usual way. Then use square brackets to deduct the marks.
- 9. Application of the marking scheme**
Apply the marking scheme as agreed.
Assistant Examiners should enter marks in Examiner Column 1.
Column 2 to be used by Advising Examiners.
Disallowed marks should be placed in square brackets i.e. '[]'.
- 10. Transfer of marks**
All marks should be transferred to the grid on the cover page of the examination answer-booklet.
Marks should be totalled, the bonus for answering through Irish applied where relevant.
- 11. Do not enter the grade for the examination in the grid on the front of the answerbook.** Enter the grade on the top right hand of the answerbook.

Junior Certificate Examination

SCIENCE

Higher Level Paper

WRITTEN EXAMINATION PAPER

Three Sections: Biology, Chemistry and Physics, *all* questions to be answered by candidates.

Biology Question 1 (52 marks); Question 2 (39 marks); Question 3 (39 marks)

Chemistry Question 4 (52 marks); Question 5 (39 marks); Question 6 (39 marks)

Physics Question 7 (52 marks); Question 8 (39 marks); Question 9 (39 marks)

COURSEWORK A

Count the number of mandatory biology investigations/experiments claimed on page 5 of the Coursework booklet and enter it in the Coursework A grid on the cover page.

Count the number of mandatory chemistry investigations/experiments claimed on page 6 of the Coursework booklet and enter it in the Coursework A grid on the cover page.

Count the number of mandatory physics investigations/experiments claimed on page 7 of the Coursework booklet and enter it in the Coursework A grid on the cover page.

Total the number of investigations / experiments claimed and award 2 marks per investigation / experiment to an amount not exceeding maximum 60 marks.

COURSEWORK B

Mark the SEC nominated investigations according to the agreed criteria. Enter the marks for each section in the Coursework B grid on the cover page of the coursework booklet.

or

Mark the candidate nominated investigation according to the agreed criteria. Enter the marks for each section in the Coursework B grid on the cover page of the coursework booklet.

Transfer of marks

Marks awarded to Coursework A and Coursework B should be transferred to the marking grid on the front of the examination paper.

SCIENCE (REVISED SYLLABUS) HIGHER LEVEL 2007
Summary of Marking Scheme

BIOLOGY

- Question 1 $(7 \times 6 + 1 \times 10)$
- Question 2 (a) $(3 \times 3), (4 \times 3)$
 (b) $(3) (2 \times 3), (3) (2 \times 3)$
- Question 3 (a) $(2 \times 3), (3 \times 3)$
 (b) $(3), (3), (3)$
 (c) $(3), (3), (3), (6)$

CHEMISTRY

- Question 4 $(7 \times 6 + 1 \times 10)$
- Question 5 (a) $(3), (3 \times 3), (3), (3)$
 (b) (4×3)
 (c) $(2 \times 3), (3)$
- Question 6 (a) $(3), (3), (3), (3), (2 \times 3)$
 (b) $(4 \times 3), (3)$
 (c) $(3), (3)$

PHYSICS

- Question 7 $(7 \times 6 + 1 \times 10)$
- Question 8 (a) $(2 \times 3), (3)$
 (b) $(2 \times 3), (2 \times 3),$
 (c) $(2 \times 3), (2 \times 3), (2 \times 3)$
- Question 9 (a) $(3), (3), (3), (3)$
 (b) $(3), (3), (6, 3)$
 (c) (4×3)

Biology (130 MARKS)
Answer each of the questions 1, 2 and 3.

Question 1. (52 Marks) All Items, (a), (b), (c), etc. (7 × 6 + 1 × 10marks)

- | | | | |
|-----|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-------------------|-------------|
| (a) | any two from: water/ salt/ urea... | (2 × 3) | [6] |
| (b) | motor function, any one from: carry messages from the brain(spinal cord) (CNS)/ to muscles (effectors) (glands)
sensory function, any one from: carry messages to the brain(spinal cord) (CNS)/ from sensory organs (sensors) (ear) (eye) (nose) (skin) tongue | (3) | [6] |
| (c) | fused/ fixed/ immovable/
no movement | (3)
(3) | [6] |
| (d) | A: lens
any one from: controls (changes) (alters) the shape (thickness) of the lens/
focuses | (3)
(3) | [6] |
| (e) | any one from: nucleus/ chromosomes/ genes/ mitochondria/ plasitds
protein | (3)
(3) | [6] |
| (f) | put tube B over the small animal
suck tube A | (3)
(3) | [6] |
| (g) | any two from production of: bread/ beer (lager) (stout)/ miso/ vinegar/
soy sauce/ wine/yoghurt/ antibiotics (penicillin)/ hormones (insulin)/
vaccines/ interferon/ biological detergents/ pesticides/ GM (genetically
modified organisms)/ bio-fuels (methane).../ spirits (vodka)... | (2 × 3) | [6] |
| (h) | Fat
test: rub food onto paper
translucent (greasy) spot
note: if a wrong food type is given by the candidate e.g. ‘starch’ then they
lose the first 4 marks. If they then give the correct test for the food that they
have named e.g. ‘iodine’ and ‘blue-black’ award the (2 × 3) | (4)
(3)
(3) | [10] |

Question 3. (39 marks) All items, (a) and (b).

- (a) (i) Complete **missing reactant:** oxygen (O₂) (3)
 missing product: carbon dioxide (CO₂) (3) **[6]**
- (ii) State **test for CO₂:** CO₂ (carbon dioxide) (3)
 lime water (3)
 goes milky (3)
 or **or**
 test for water: cobalt chloride/ copper sulphate (3)
 add water (3)
 turns pink/ turns blue (3) **[9]**
 note: colour must be matched with reagent
- note:** if the candidate gives oxygen, O₂, as a **product** in (i) they get no marks for it in (i), **allow** in (ii) (3) for ‘relights’ and (3) for ‘glowing splint’ i.e. in this case test for O₂ can get (2 × 3)
- (b) Name oxygen (O₂) (3) **[3]**
- Name carbon dioxide (CO₂) (3) **[3]**
- How? brighter (more) light/ increase (more) CO₂ (3) **[3]**
- (c) (i) Explain burning releases carbon dioxide (CO₂) (3)
 or **or**
 any one from: deforestation results in less carbon dioxide (CO₂) being used/ less photosynthesis/ burning forests releases CO₂ (3) **[3]**
- (ii) Suggest **any one from:** acidification of rivers (lakes) (seas) / melting of polar ice (glaciers)/ rise in sea levels / drought/ greenhouse effect/ disruption of aquatic food chains/ climate change (hotter) (colder) (more or less rain) (more or less wind)/ changes in ocean currents/ extinction of species/ global warming... (3) **[3]**
 note: ‘damages the ozone layer’ gets no marks
- (iii) Suggest **any one from:** carbon dioxide (CO₂) is taken in by plants/ used in photosynthesis/ more leaves/ less fuel is burnt for heating... (3) **[3]**
- (iv) How? **Allow any one from:** plant more trees (reforestation)/ by increasing photosynthesis/ by increasing photosynthesis/ burn less fuel/ turn down thermostats/ better heat insulation of houses (correct example)/ heat pump/ bio-fuel/ solar panels... (6) **[6]**
 note allow 6 marks for ‘burn less fuel only it has not appeared in (iii) above

Chemistry (130 MARKS)
Answer each of the questions 4, 5 and 6.

Question 4. (52 marks) All items, (a), (b), (c), etc. (7 × 6 + 1 × 10marks)

- (a) two dots (Xs) in inner circle and eight dots (Xs) in ‘middle’ circle (3)
 one dot (X) in outer circle (3) [6]
- (b) oil (3)
any one from: will not rot (decay)/ bacteria (fungi) (microbes) (micro-organisms) cannot break them down (3) [6]
accept: can not be decomposed (broken down) for 3 marks
- (c) graduated (measuring) cylinder (3)
any one from: burette/ pipette/ gas (graduated) syringe (3) [6]
- (d) **any two from:** soft (can be cut with a knife)/ low density(float on water)/ shiny (lustrous) when cut/ tarnish(form oxide) (reacts) with air/ burn in air/ coloured flames/ react with water/ hydrogen produced with water/ hydroxides formed with water/ silver’ (white metals)/ very reactive/ one electron in outer orbit... (2 × 3) [6]
- note** general properties of most metals like electrical and thermal conductivity, ductility, malleability, etc get no marks, specific properties of **alkali** metals are required.
- (e) shared (3)
 electrons (3) [6]
- (f) **any one from:** alum/ copper sulphate/ salt/ sugar/ silicon/ iodine/ diamond... (3)
any one from: crystalline solids have definite (geometric) shapes/ particles in crystals are in regular(geometric) order/ texture e.g. salt and sugar ‘feel’ different to flour/ crystals affect light/ shiny / sparkle... (3) [6]
- (g) **any one from:** Ca(OH)₂/ CaCO₃/ NaOH/ Na₂CO₃/ NaHCO₃/ NH₃... (3)
any one from: sodium hydroxide (caustic soda)/ sodium carbonate (washing soda) sodium hydrogen (bi) carbonate (bread soda)/ ammonia... (3)
accept any one from: the named household substances: toothpaste/ oven cleaner/ antacid (named antacid)/ baking powder... (3) [6]
do not accept: bleach/ shampoo
note: accept names **only** for the second 3 marks
- (h) A condenser (3)
 Y (3)
 Water/ H₂O (2)
 salt/ sodium chloride/ NaCl... (2) [10]

Question 6. (39 marks) All items, (a), (b) and (c).

- (a) Give HCl, **accept** formulae(s) of other acids (3) [3]
- Give calcium carbonate
accept any named carbonate **or** bi(hydrogen) carbonate (3) [3]
- What? denser (heavier) than air (3) [3]
- What? both red (pink)/ blue to red (pink) (3) [3]
- Give **any two from:** fire extinguishers/ fizzy drinks/
photosynthesis/ 'dry ice'/ 'stage effects'... (2 × 3) [6]
- (b) (i) Describe **test:** add soap (3)
shake (3)
result: any one from: water in flask **A** (hard water) does
not form a lather easily with soap / forms a scum with soap (3)
any one from: flask **B** contains water that forms a
lather easily with soap/ does not form a scum
with soap/ contains soft water (3) [12]
- (ii) What? **any one from:** compounds of calcium
(magnesium)/ calcium **or** magnesium ions (salts)/
formulae(s) **or** names of compounds of calcium
and magnesium, **except carbonates**, e.g. CaCl₂,
Ca(HCO₃)₂, calcium chloride, calcium hydrogen
carbonate.../ lime (3) [3]
- (c) (i) Name **any one from:** screening/ settling/ filtration/ UV
'light'/ chlorination/ fluoridation/ pH adjustment/
softening/ distillation... (3) [3]
- (ii) Give **any one from (matched):**
Screening: takes out large solids
settling: solids sink
filtration: solids are removed
UV (ultraviolet) irradiation: stops certain
dangerous microbes multiplying
chlorination: to kill bacteria (microbes)
fluoridation: help prevent tooth decay
pH adjustment: stop damage to metal pipes
softening: removing dissolved calcium/
magnesium
distillation: removing dissolved solids (3) [3]

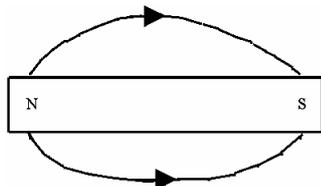
Physics (130 MARKS)
Answer each of the questions 7, 8 and 9.

Question 7. (52 marks) All items, (a), (b), (c), etc. (7 × 6 + 1 × 10marks)

- (a) (i) heat the ball, it does not pass through the ring (3)
(ii) let the ball cool, now passes through the ring (3) [6]
- (b) (i) ice is less dense than water (3)
(ii) ice is more dense than ethanol (3) [6]
- (c) bulb A lights (3)
the diode with A is in forward bias (allows current to flow)
(+ end of diode connected to + pole of battery) (3) [6]
note allow 3 marks for a correct reason for why B does not
light if a correct reason for why A does light is not given
- (d) **magnetic effect any one from:** electromagnets/ door bells/ remote opening of
doors/central locking in cars/ phones/ speakers/ electric motors/ any named
device incorporating an electric motor/ read (write) heads on audio (video)
recorders computer drives/ remote switching e.g. starter motor in car
electrical meters... (3)
chemical effect any one from: electroplating/ chromium plating/ silver
plating/cold galvanising/ refining of copper/ refining of aluminium/
production of sodium hydroxide/ production of hydrogen/
charging a battery... (3) [6]
accept 'electrolysis' for 3 marks
- (e) light moves faster than sound (6) [6]
- (f) **advantage any one from:** no 'greenhouse gas 'emissions/ no carbon dioxide
(CO₂) emissions/ supply of nuclear fuel for many centuries/
can produce large amounts of energy/ cleaner supply of electricity... (3)
disadvantage any one from: devastation if a reactor erupts/
very long term storage of wastes/ wastes are radioactive(dangerous)/
mining for nuclear fuel damages the environment /
transport of nuclear fuel is hazardous/ terrorist threat
reprocessing nuclear fuel is hazardous (can damage the environment)/... (3) [6]
- (g) water is a poor conductor of heat (3)
any one from: hot water rises/ water is heated by convection(current) (3) [6]
note 'heat rises' alone gets no marks
- (h) **weather conditions:**
any two from: cloudy/ windy/ rain... (2 × 3)
explanation:
any one from: air rises/ water vapour condenses (cools)/
air moves in (4) [10]

Question 8. (39 marks) All items, (a), (b), (c), etc.

(a) Draw



two magnetic field lines shown, **one on each side** of the bar magnet. **[if direction is not given or given incorrectly for a magnetic field line award no marks for that 'line']**

(2 × 3) [6]

What?

poles/ north and south

(3) [3]

(b) Calculate

note
 (i) if no calculation is performed allow 3 marks for 'work = force × distance', award no marks for units in this case.
 (ii) allow 2 marks for (8 × 20) or (8000 × 20), if the units are matched allow the 3 marks for the unit
 (iii) unit alone gets no marks

160 000

(3)

J (Nm)

(3)

or
160

(3)

or
kJ (kNm)

(3)

[6]

Identify

any one from: kinetic to heat/ kinetic to sound/ kinetic to electrical/ kinetic to chemical/ kinetic to potential
note: the last three are included to allow for 'hybrid' cars
[allow 3 marks for a correctly named energy provided it is on the correct side of 'to' or of an arrow]

(2 × 3) [6]

(c) Define

accept turning effect or turning power for 3 marks

force (weight) multiplied by its distance from the fulcrum

(3)
(3)

[6]

Calculate

$30 \times F = 40 \times 3$
 $F = 4 \text{ N}$ ('N' not required)
note (i) allow 5 marks for $(40 \times 3)/30$ if it is the only statement
 (ii) '4' alone merits 6 marks

(3)
(3)

[6]

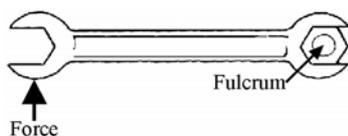
Give

e.g. spanner and nut
 accept **any** correct **everyday** example of a lever

fulcrum shown correctly
 one force shown correctly
[no diagram deduct 3 marks]

(3)
(3)

[6]



note: if fulcrum **and** force are not shown in the diagram allow (3) for a correct example of an application of a lever shown

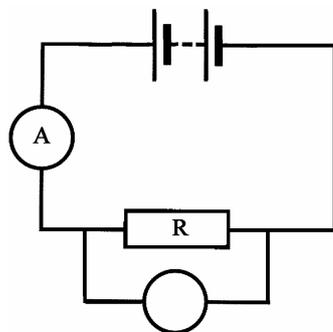
note if no diagram is given allow 3 marks for **any one from:** correct named application e.g. 'door'/ correct fulcrum named e.g. 'hinge'/ correct position of a force e.g. 'handle'

Question 9. (39 marks) All items, (a) and (b).

- (a) (i) What? refraction (3) [3]
(ii) Pick **R** (3) [3]
(iii) Give **any one from:** lenses/ spectacles/ magnifying glass/
microscope/ binoculars/ telescopes/ camera lenses/
prisms/ projectors/ dispersion (rainbow) (spectrum)/ dioscope/
endoscope/ periscope/... (3) [3]
(iv) Name **any one from:** reflection (bouncing off) (mirror)/ gravity (3) [3]

- (b) What? current/ Amperes (Amps) (3) [3]

Enter



A correctly shown in series with **R** and the battery (3) [3]

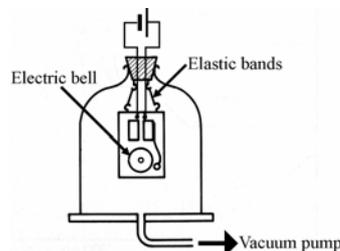
Use

12, **accept** 11.5 to 12.5 i.e. +/- 0.5
 Ω or Ohms (6)

allow 3 any correct ratio from the graph e.g. $\frac{1.2}{0.1}$ or $\frac{1.8}{0.15}$ etc (3) [9]

note if an incorrect ratio is given but it is used correctly in a calculation allow 3 marks e.g. $\frac{6}{5} = 1.2$

- (c) Describe



show or state

- battery/ cell/ source of electricity (3)
bell in jar (container) (3)
vacuum pump/ air removed (3)
bell is less loud (can't be heard) (3) [12]
[no diagram deduct 3 marks]

acceptable sources of sound include: mobile phone, alarm clock, kitchen timer. Look for an equivalent point e.g. 'dial' for the phone for the first 3 marks

[if the candidate does not score 12 marks for the above but has 'elastic bands'/ sound insulation between bell and container award 3marks]
accept equivalent experiments, look four equivalent points

PHYSICS – Marking Criteria for Coursework B

			Guide to mark assignment	
Section	Aims	Total Mark	Investigate the relationship between the length of a metallic conductor and its resistance	H.L.
Introduction	Clear statement of the problem/topic to be investigated, background research undertaken in preparation for the investigation: people, books, websites, etc. as sources of relevant information.	5	Statement / identification of problem / topic to be investigated:	(2)
			Research: Any reference to book / web / person consulted etc.	(3)
Preparation and planning	Identification of variables and controls as required	20	Variables / Controls: Identify any <i>four</i> variables and/or indicate how some of these need to be controlled or held fixed: Material of the wire / Cross-sectional area (thickness) / Length / Temperature / Resistance / Extended wire but not stretched / Same multimeter (leads)	(3 + 3 + 2 + 2)
	List of equipment needed for the investigation		Equipment needed: Identify any <i>three</i> pieces of equipment used: wire / metre stick / ohm meter (multimeter) / thermometer / voltmeter / ammeter / metre bridge / wheatstone bridge / leads / crocodile clips / wire cutters / Rheostat	(1 + 2 + 2)
	List of tasks to be carried out during the investigation		List of tasks: Identify any four tasks carried out in investigation: measuring (cutting) length of wire / measuring resistance / varying length / recording resistance and length in a table calculations / graphing	(2 + 1 + 1 + 1)

CHEMISTRY – Marking Criteria for Coursework B

		<u>Guide to mark assignment</u>		
Section	Aims	Total Mark	H.L.	
			Investigate how the conc. of a H ₂ O ₂ solution affects the speed at which it decomposes to produce oxygen gas	
Introduction	Clear statement of the problem/topic to be investigated, background research undertaken in preparation for the investigation: people, books, websites, etc. as sources of relevant information.	5	Statement / identification of problem / topic to be investigated:	(2)
			Research: Any reference to book / web / person consulted etc	(3)
Preparation and planning	Identification of variables and controls as required	20	Variables / Controls : Identify any <i>four</i> variables and/or how some of these can controlled or held fixed: concentration of peroxide / volume of peroxide solution / amount of catalyst / particulate size of catalyst / activity of catalyst / temperature / rate of reaction (volume over time)	(3 + 3 + 2 + 2)
	List of equipment needed for the investigation		Equipment needed: Identify any <i>three</i> pieces of equipment used: buchner flask (reaction flask) and delivery tube / stopper / bee-hive shelf / graduated cylinder / trough of water / gas syringe / water bath / H ₂ O ₂ solution / MnO ₂ (celery) / Same catalyst / thermometer / washing up liquid	(1 + 2 + 2)
	List of tasks to be carried out during the investigation		List of tasks: Identify any <i>four</i> tasks carried out in investigation: vary concentration / weigh catalyst / measuring rates / prepare catalyst in suitable manner / calculation / record results / graph	(2 + 1 + 1 + 1)

BIOLOGY – Marking Criteria for Coursework B

			Guide to mark assignment	
Section	Aims	Total Mark	Quantitative survey of the plant species in a local habitat	H.L.
Introduction	Clear statement of the problem/topic to be investigated, background research undertaken in preparation for the investigation: people, books, websites, etc. as sources of relevant information.	5	Statement / identification of problem / topic to be investigated:	(2)
			Research: Any reference to book / web / person consulted etc.	(3)
Preparation and planning	Identification of variables and controls as required	20	Variables / Controls: Investigation doesn't involve normal variables/controls // or Identify any one factors which contributed to a kept fair: quadrat size / habitat / distance between transect intervals / randomness / on one visit	(5)
	List of equipment needed for the investigation		Equipment needed: Identify any three pieces of equipment used: quadrats / transect / equipment used to measure area / key / notepad (clipboard) / something to throw (for randomness)	(1 + 2 + 3)
	List of tasks to be carried out during the investigation		List of tasks: Identify any four tasks carried out in investigation: choose a habitat / throw pen (quadrat) randomly (do quadrat study) / set out transect (do transect study) / area measurement / identify / count / note (record data) / repeat what's to be measured / calculation / graph	(3 + 2 + 2 + 2)

OWN INVESTIGATION – Marking Criteria for Coursework B

Guide to mark assignment				
Section	Aims		Total Mark	H.L.
Introduction	Clear statement of the problem/topic to be investigated, background research undertaken in preparation for the investigation: people, books, websites, etc. as sources of relevant information.	10	Statement / identification of problem / hypothesis statement / topic to be investigated: Research: Any <i>two</i> references to book / web / person consulted etc	(6) (2 × 2)
Preparation and planning	Identification of variables and controls List of equipment needed for the investigation List of tasks to be carried out during the investigation	40	Variables & Controls*: Identify any <i>four</i> variables / controls: Equipment needed: Identify any <i>five</i> pieces of equipment used List of tasks: Identify any <i>four</i> tasks carried out in investigation If variables/controls not relevant to the type of investigation undertaken allow 10 marks for stating so and then readjust equipment to (3 + 3 + 4 + 5) <u>and</u> tasks to (3 + 3 + 4 + 5)	(4 × 5) (5 × 2) (2 + 2 + 3 + 3)
Procedure	Procedure, apparatus, safety, data collection/observations <ul style="list-style-type: none"> ▪ Safety precautions required for this investigation ▪ Procedures followed in the investigation ▪ Recorded data/observations 	40	Safety: Identify any <i>two</i> safety precaution followed in conducting the investigation Procedure: State <u>or</u> Show Identify any <i>eight</i> steps taken in conducting investigation Recorded Data / Observations: Identify any <i>two</i> points related to method used	(2 × 3) (8 × 3) (2 × 5)
Analysis & Conclusions	Analysis <ul style="list-style-type: none"> ▪ Calculations/data analysis ▪ Conclusion(s) and evaluation of results(s) 	40	Calculations / Data analysis: <i>Two</i> relevant comment analysing data or calculation or graph Limited manipulation of data OR Good manipulation of data Conclusion: <i>Two</i> relevant conclusion drawn or evaluation of results obtained Limited treatment OR Good treatment	(4) } (7) } × 2 (10) } (4) } (7) } × 2 (10) }
Comment	Comments (e.g. refinements, extensions, sources of error etc.)	20	Three comments on refinements / extensions / sources of error e.g. What was learnt** / reliability of data / how process could be improved / sources of error / possible reason for unexpected result	(5 + 5 + 10)
** Other than the conclusions already stated				